Analysis of Cation Migration Channels in PbWO₄

Volodymyr Shevchuk, Ihor Kayun Department of Electronics and Computer Technologies, Ivan Franko National University of Lviv Drahomanova St 50, 79005 Lviv, Ukraine e-mail: <u>shevchuk@electronics.lnu.edu.ua</u>

Abstract—The analysis of possible W-ions migration in $PbWO_4$ with sheelite type structure was realized using the TOPOS calculation complex program. The temperature dependence of migration channel lengths and theirs evolution with change of unit cell parameters were considered.

Index Terms—PbWO₄; sheelite; electromigration; TOPOS; W-migration path; migration channel

I. INTRODUCTION

The analysis method of crystalline space and totality of voids and channels in crystal structure based on Voronoi tessellation was realized in TOPOS program package [1] for ionic electrical migration in different crystals. The topological atomic analysis, visualization of cationic migration path, and some basic concepts on micro-level investigation of the mobile ion migrations in crystal lattice using TOPOS calculation complex programs was carried out for real oxide crystals in particular with sheelite type structure in our works [2, 3].

In the present paper with the aim to analysis of microstructure and features of the migration ways of W-ions in $PbWO_4$ were considered a numerous literature data and our experimental results on X-ray crystallographic investigations. We calculated and constructed the W-migration path in compounds and considered a temperature changes of characteristic elementary parameters of cationic transference specifically the length of elementary channels. The evolution of the calculated migration channel lengths was analyzed with changing of a unit cell parameters. Consideration of chemical and structural factors and visualization of conduction pathways are promoted the determination of mechanisms and other spatial features of ions migration in crystals. This approach allowed also the forecast of crystal properties and applications.

II. EXPERIMENTAL AND CALCULATION PROCEDURE

Our PbWO₄ undoped single crystal (sheelite type structure, space group $C_{4h}^6 - I4_1/a$) was grown by Czochralski technique [4]. Other PbWO₄ compounds (sheelite type structure) were prepared by a different authors and methods (see Table I). The structural data and densities of PbWO₄ (Table I) obtained from the X-ray diffraction (XRD) data. Our XRD analysis was performed on a STOE STADI P powder diffraction system. Arrays of experimental intensities and diffraction angles were obtained using the mentioned diffractometer equipped with a linear position sensitive detector PSD in a modified Guinier geometry scheme in Bragg-Brentano transmission mode. Conditions of the measurements: monochromatic CuK α_1 radiation (λ =1.540598 Å); bent Johann type [111] Gemonochromator; $\omega/2\theta$ -scan; 2 θ -range 4° $\leq 2\theta \leq 120$ °; step 0.480° (2 θ); step scan time 250 s. Other XRD-measurement details were described previously [15]. The search for structural data was performed using *Pearson's Crystal Data* [16].

In the work using the program package TOPOS 4.0 the Wionic migration maps for the $PbWO_4$ crystals were constructed. Possible migration paths of W ions in the structure crystals at different temperatures (ref. [14]) and changing of a unit cell volume (data in the Table I) were analyzed.

TABLE I. Some Crystallographic and Physical Data for the $PBWO_4$ Compounds used for Calculations

Ref	Method of	Unit cell param., Å		Unit cell	Density,
	growing	а	С	volume, Å ³	g cm ⁻³
Our data	Czochralski	5.4601	12.0425	359.02	9.0662
[5]	Czochralski	5.4560	12.02	357.81	9.0968
[6]	Czochralski	5.4360	11.9570	353.33	9.2122
[7]	mechanical alloying	5.4661	1.20779	360.87	9.0198
[8]	conventional solst. reaction	5.4597	12.0420	358.95	9.0679
[9]	Czochralski?	5.4646	12.0479	359.77	8.8177
[10]	natural	5.4450	12.0495	357.24	9.1113
[11]	complex polymerization	5.4637	12.0654	360.18	9.0371
[12]	precipitation method	5.4645	12.0553	359.98	9.0420
[13]	polycrystal	5.3851	11.7223	339.94	9.5751
[14]	Czochralski	5.4632	12.0482	359.60	9.0516

The map of voids and channels is consistent with experimental data. The basic concepts for description of the voids and channel are the following: elementary void (channel), and closely related terms of form and radius of void, significant elementary void (channel). The Voronoi polyhedron of atom (geometric image atom), as suggested by the value of the second moment inertia (G).

Elementary crystal void is an area of crystal unit cell, the center of which is one of the vertexes of Voronoi polyhedron. The major (ZA) and minority (ZC) elementary voids with sequence numbers N (ZAN and ZCN) are considered. To show preference for a ZAN voids. For the calculation we used the radii of the ion W^{6+} (0.56 Å) at the coordination number 4, Pb^{2+} (1.43 Å) at the coordination number 8, and O^{2-} (1.36 Å) [17]. For W (as near to Mo) the value of the parameter G is

0.0830(1) [18]. The migration way is determined as the set of elementary voids and lines of elementary channels. It can be infinite along the 1D-, or 2D, or 3D-channel-net. The conductivity map formed by the migration pathways of possible charge carriers.

III. RESULTS AND DISCUSSION

Analysis of the calculation data in most cases the continuous pathways for the Pb- and W-ions in the PbWO₄ compound with sheelite type structure were not observed. But for some crystal (structure data [11, 19]) the continuous pathways for the W-ions were observed (Fig. 1). The technological condition of compound preparation play the principal role in considered cases and determined the variation unit cell volumes and void distances (Fig. 2) correspondingly. The great meaning of unit cell volume stimulate formation the continuous ways for the W-ions probable migrations.



Figure 1. Probable migration pathways calculated at RT for the W ions in PbWO₄ using crystal data [11] and fragment (inset) of channel migrations.



Figure 2. Evolution of some voids distances at RT with changing of a unit cell volume for the $PbWO_4$ prepared by different methods (see Table 1).

The Fig. 3 (a, b, c) show the temperature dependencies of the distances between ZAN-voids in possible migration channels of W-ions. The calculations were performed for X-ray crystal structure data [14].

The curves on Fig. 3 demonstrate specific temperatures about 100 and 200 K. At these temperatures the change of the declination curves (change of the temperature dependence law) are observed. Up to the temperature 100 K the rising of channel length are not observed. In temperature range 100-200 K are occur the nearly linear increasing/decreasing of the channel lengths (links or the same as distances between neighboring voids). These features of temperature behavior changing of the distance voids in PbWO₄ are connectable to changing of other physical and structural [14] properties of crystal at above-mentioned temperatures.



Figure 3. Temperature dependencies (a, b, c) of the some void distances. Calculation was performed using data [14].

The calculation data of a ZAN voids and suitable connections for some crystals are presented in the Table II.

Ref	ZAN	Positions			Void bound	Length,
		х	у	z	volu boullu	Å
Our data	ZA1	0.87586	0.31459	0.76202	ZA1-ZA4	0.774
	7.4.2	0.68489	0.11120	0.66520	ZA3-ZA4	0.318
	ZA3				ZA3-ZA5	0.577
	ZA4	0.67351	0.15972	0.67883	ZA4-ZA1	0.774
					ZA4-ZA3	0.318
	ZA5	0.63829	0.17805	0.63473	ZA5-ZA3	0.577
					ZA5-ZA4	0.574
	ZA6	0.96585	0.68436	0.65422	ZA6-ZA7	0.419
	ZA7	0.00000	0.75000	0.66330	ZA7-ZA6	0.420
	ZA8	0.04775	0.77717	0.67051	ZA8-ZA6	0.297
					ZA8-ZA9	1.721
					ZA8-ZA9	1.832
	ZA9	0.40656	0.91419	0.56001	ZA9-ZA8	1.832
					ZA9-ZA8	1.721
	ZA1	0.88833	0.31673	0.75716	ZA1-ZA4	0.974
	7.1.2	0 (7071	0.11000	0.66192	ZA3-ZA4	0.172
	ZA3	0.6/8/1	0.11980		ZA3-ZA8	1.590
	7.1.4	0.67298	0.14663	0.66884	ZA4-ZA1	0.974
	ZA4				ZA4-ZA3	0.172
		0.65228	0.15900	0.64312	ZA5-ZA4	0.337
	ZAS				ZA5-ZA9	0.912
[11]	ZA7	0.04524	0.77524	0.67696	ZA7-ZA8	1.742
					ZA7-ZA8	1.755
					ZA7-ZA9	0.247
	ZA8	0.34485	0.66345	0.69036	ZA8-ZA3	1.590
					ZA8-ZA7	1.742
					ZA8-ZA10	1.251
	ZA9	0.96631	0.69363	0.66302	ZA9-ZA5	0.912
					ZA9-ZA7	0.247
	ZA10	0.50000	0.75000	0.62500	ZA10-ZA8	1.251
	ZA1	0.89058	0.31376	0.76319	ZA1-ZA4	0.880
	ZA3	0.66678	0.13115	0.66200	ZA3-ZA9	1.600
	ZA4	0.66142	0.15593	0.66900	ZA4-ZA1	0.880
	ZA5	0.64307	0.16663	0.64698	ZA5-ZA6	0.879
	ZA6	0.97347	0.70447	0.65512	ZA6-ZA5	0.879
					ZA6-ZA8	0.187
[10]	ZA8	0.03452	0.77155	0.66553	ZA8-ZA6	0.187
[19]					ZA8-ZA9	1.852
					ZA8-ZA9	1.945
	ZA9	0.40691	0.89911	0.56495	ZA9-ZA10	1.202
					ZA9-ZA3	1.600
					ZA9-ZA8	1.852
					ZA9-ZA8	1.945
	ZA10	0.50000	0.75000	0.62500	ZA10-ZA9	1.202

TABLE II. Position of calculation voids and channel lengths for W-migration at RT in the \mbox{PBWO}_4 with sheelite type structure

Analysis of the results in most cases shows realization the unlinked channels W-migration. The probable continuous migration way of W-ions (along [001]-axis) in picked crystals [11, 19] at RT carried out through "intermediate" voids what spaciously are not equivalent to regular positions of W-ions.

IV. CONCLUSIONS

The possible migration ways for the W-ions in PbWO₄ were visualized using the program package TOPOS. For some structural data at RT a possible continuous 3D-network of W-ions migration is formed. The shape of the migration way and the void bounds are determined by method and technological

conditions compound preparation and by temperature of investigated samples.

REFERENCES

- V. A. Blatov, "Multipurpose crystallochemical analysis with the program package TOPOS", IUCrCompCommNewsLetter, iss. 7, pp. 4-38, 2006.
- [2] V. N. Shevchuk and I. V. Kayun, "Migration ways of ions in CaMoO₄ and BaWO₄ crystals with sheelite-type structure," Proc. Int. Conf. Oxide Materials for Electronic Engineering – fabrication, properties and applications OMEE-2014, Ukraine, pp. 117-118, May, 2014.
- [3] V. N. Shevchuk and I. V. Kayun, "Analysis of electromigration and structure of AWO₄ (A=Ca, Cd, Pb, Zn) crystals using the TOPOS program complex", Chem. Met. Alloys, vol. 9, pp. 128–134, December 2016.
- [4] V. N. Shevchuk and I. V. Kayun, "Influence of thermal prehistory on the electrical properties of tungstate crystals", Chem. Met. Alloys, vol. 4, pp. 72–76, June 2011.
- [5] J.M. Moreau, Ph. Galez, J.P. Peigneux, and M. V. Korzhik, "Structural characterization of PbWO₄ and related new phase Pb₇W₈O_(32-x)," J. Alloys and Comp., vol. 238, pp. 46–48, May 1996.
- [6] D. Errandonea, J. Pellicer-Porres, F. J. Manjón, A. Segura, Ch. Ferrer-Roca, R. S. Kumar, O. Tschauner, J. López-Solano, P. Rodríguez-Hernández, S. Radescu, A. Mujica, A. Muñoz, and G. Aquilanti, "Determination of the high-pressure crystal structure of BaWO₄ and PbWO₄," Phys. Rev. B, vol. 73, pp. 224103-1-224103-15, June 2006.
- [7] Sh. Takai, T. Nakanishi, K. Oikawa, S. Torii, A. Hoshikawa, T. Kamiyama, and T. Esaka, "Neutron diffraction and IR spectroscopy on mechanically alloyed La-substituted PbWO₄," Sol. St. Ionics, vol. 170, pp. 297-304, May 2004.
- [8] Sh. Takai, Sh. Touda, K. Oikawa, K. Mori, Sh. Torii, T. Kamiyama, and T. Esaka, "Powder neutron diffraction study of Ln-substituted PbWO₄ oxide ion conductors," Sol. St. Ionics, vol. 148, pp. 123-133, June 2002.
- [9] R. Chipaux , G. Andre, and A. Cousson, "Crystal structure of lead tungstate at 1.4 and 300 K," J. Alloys and Comp., vol. 325, pp. 91–94, July 2001.
- [10] Xu Kegin, Xue Jiyue, and Ding Yang, "Discovery of stolzite in China and refinement of its crystal structure," Acta Geologica Sinica, vol. 8, pp. 111-116, March 1995.
- [11] L.S. Cavalcante, J.C. Sczancoski, V.C. Albarici, J.M.E. Matos, J.A. Varela, and E. Longo, "Synthesis, characterization, structural refinement and optical absorption behavior of PbWO₄ powders," Materials Science and Engineering B, vol. 150, pp. 18-25, April 2008.
- [12] R. Saraf, C. Shivakumara, S. Behera, H. Nagabhushana, and N. Dhananjaya, "Facile synthesis of PbWO₄: Applications in photoluminescence and photocatalytic degradation of organic dyes under visible light," Spectrochimica Acta Part A: Molecular and Biomolecular Spectroscopy, vol. 136, pp. 348-355, February 2015.
- [13] A. Grzechnik, W. A Crichton, W. G Marshall, and K. Friese, "Highpressure x-ray and neutron powder diffraction study of PbWO4 and BaWO4 scheelites," J. Phys.: Condens. Matter., vol. 18, pp. 3017-3029, March 2006.
- [14] D. M. Trots, A. Senyshyn, and B. C. Shwarz, "Low temperature structural variation and heat capacity of stolcite PbWO₄", J. Sol. St. Chem., vol. 183, pp. 1245-1251, June 2010.
- [15] V. N. Shevchuk and I. V. Kayun, "Nano- and micro-size V_2O_5 structures", Chem. Met. Alloys, vol. 4, pp. 67–71, June 2011.
- [16] "Pearson's Crystal Data Crystal Structure Database for Inorganic Compounds", Release 2015/16, ASM International, Material Park, P. Villars and K. Cenzual, Eds., Ohio, USA.
- [17] "Moden Crystallography", B. K. Vainshtein, Ed., vol. 2. Moscow: Nauka, 1979, p. 359.
- [18] V. A. Blatov, "Voronoi-Dirichlet polyhedra in crystal chemistry: theory and applicaions", Crystallogr. Rev., vol. 10, pp. 249-318, April 2004.
- [19] R. M. Hazen, L. W. Finger, and J. W. E. Mariathasan, "High-pressure crystal chemistry of sheelite-type tungstates and molibdates", J. Phys. Chem. Solids, vol. 46, pp. 253-263, Februar 1985.